

Research Article

Synthesis and characterization of new Schiff base derived from 2-amino-5-(substituted phenyl) thiadiazole, substituted aromatic aldehyde and acetyl acetone

Article DOI:[10.55434/CJST.2022.10102](https://doi.org/10.55434/CJST.2022.10102)**Author's Affiliations**

*¹Department of Chemistry,
Dayanand Science College,
Latur- 413512,
(MS), India.*

Corresponding Author**Dr. S.N. Ibatte**Email: ibateshyam@gmail.com*Received- 17thFebruary 2022,**Accepted- 14th March 2022*

©2022 The Authors Published under
**Caribbean Journal of Science and
Technology**

ISSN 0799-3757<http://caribjstech.com/>**S. N. Ibatte****Abstract**

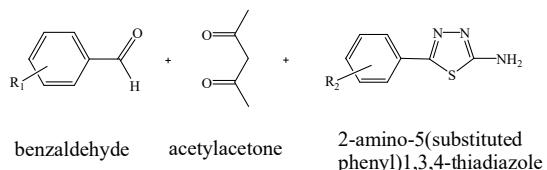
All the aimed new Schiff bases are synthesized from 2-amino-5-(substituted phenyl) thiadiazole, substituted aromatic aldehyde and acetyl acetone using silica supported TBAHS as catalyst at reflux temperature and are sufficiently characterized by IR, ¹H NMR and mass spectra.

Keywords: Thiadiazol; Acetylacetone; Aromatic aldehyde; NMR; Schiff base**Introduction**

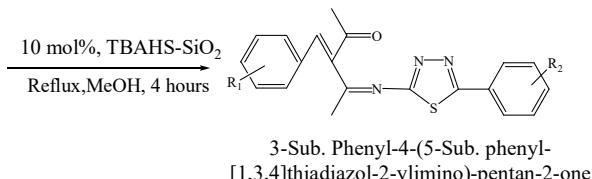
Schiff bases play crucial role in biological activity. Schiff bases containing a target molecules works in various types of activities in pharmacology. Mostly, Schiff bases containing azomethine (-C=N-) type of linkage which introduces condensation reaction of carbonyl compound such as aldehydes & ketones with primary aliphatic or aromatic amines. Schiff bases are well known for their antitumor, antifungal, antiviral, antibacterial, and anti-inflammatory activity¹⁻¹⁹. Today researchers are very much interested to make various types of heterocyclic compounds with help of universal multicomponent route which is more economic, affordable and simple one for the synthesis in one pot manner (without intermediate formation). The activity of Schiff bases were accelerated by forming metal complexes. Metal complexes have been used widely as bactericides, fungicides, insecticide and pesticides²⁰⁻²⁷. Accordingly metal complexes of Schiff bases, containing hetero atoms such as N, S, and O etc. had several biopotencies. Gram negative bacterial infections are very severe than the gram positive bacterial²⁸⁻³². Considering the higher risk for growing antibacterial resistance it is necessary to discover a unique/novel Schiff base with more potent and resistance in pharmacology. Hence, keeping in view of the importance of the Schiff bases we planned to synthesize a library of Schiff bases from 1,3,4-thiadiazole moiety, substituted aldehydes and di-ketones. The derived products were spectroscopically analyzed and also screened by antimicrobial study.

Present work:

The synthesis and characterisation of new Schiff base ligands are introduced in this paper. We present a novel approach for synthesising Schiff bases with three components in a single step. Schiff base ligand produced from 2-amino-5-(substituted phenyl)



thiadiazole, substituted aromatic aldehyde, and acetyl acetone, which was recently synthesised. The structural characteristics were determined using IR, ¹H NMR, and Mass. Antimicrobial characteristics of the produced compounds were also tested (**Scheme 1**).



Scheme 1: TBAHS-catalyzed synthesis of Schiff bases

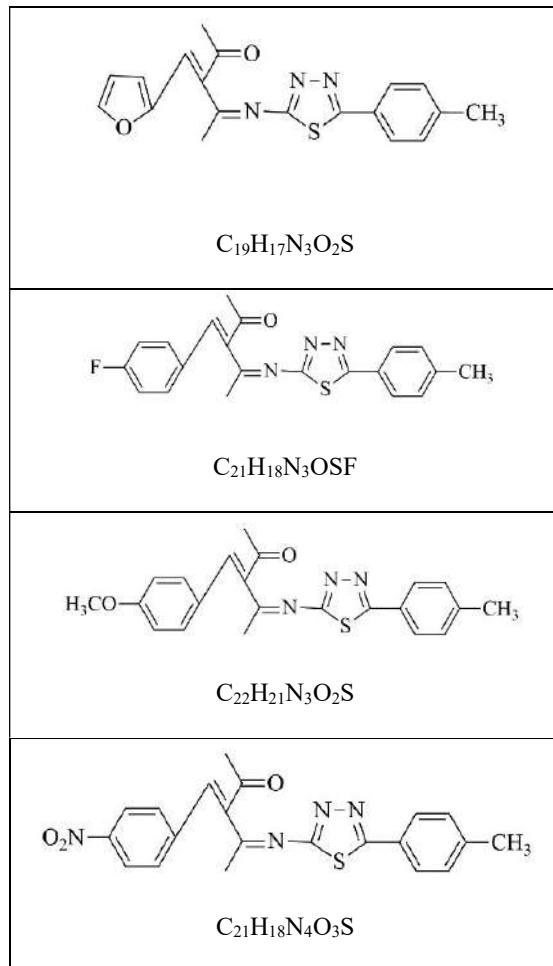
Experimental

Typical procedure of Synthesis of 3-phenyl-4-(5-phenyl-[1,3,4]thiadiazol-2-ylimino)-pentan-2-one

On a water bath, acetyl acetone (10 mmol), 2-amino-5-(substituted phenyl)thiadiazole (10 mmol), and substituted aromatic aldehyde (10 mmol) were refluxed with 10 mol percent solid supported TBAHS until the reaction was completed. Thin layer chromatography was used to monitor the reaction's completeness (TLC). The catalyst was filtered after the reaction was completed, washed with methanol, and the filtrate concentrated under reduced pressure. Recrystallization with hot methanol purified the crude product. The recovered catalyst was activated for two hours at 180°C and reused four times for complex preparation. For new reactions, the catalyst's activity remains unchanged.

Result and Discussion

Commonly, the Schiff bases were obtained by heating under reflux condensation of particular substituted aldehyde, diketone and substituted thiadiazole in 1:1:1 proportion of molar ratio³³. The coined Schiff bases purity was determined by TLC using silica gel as adsorbent and solvent system as benzene, acetone (7:3 v/v) ratio. Schiff bases gave an exact melting point showing the purity of ligand. Also molecular weight and elemental analysis of Schiff bases acquired were in good agreement with expected formulas of Schiff bases. The structural, spectroscopic and analytical data were shown as per expected formulae.



Structure & molecular formula

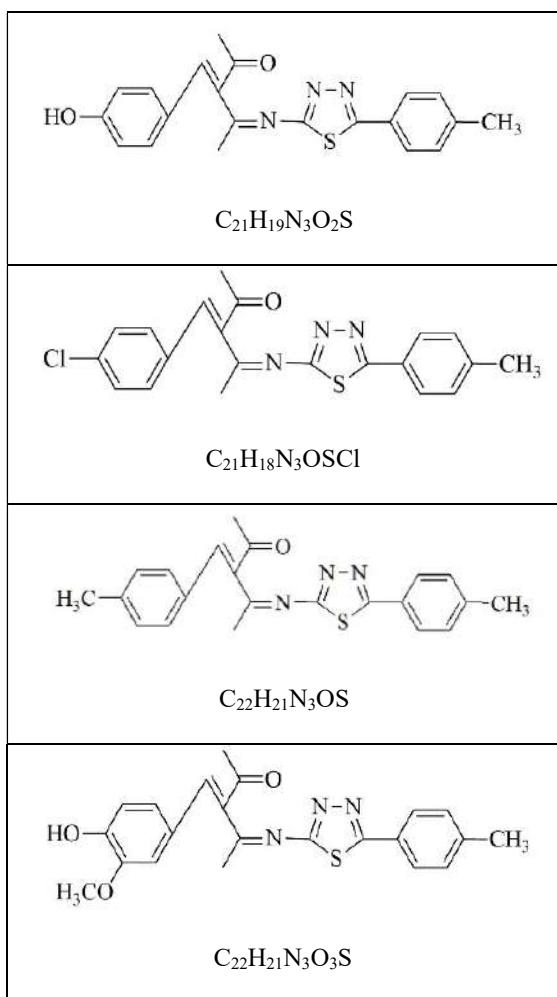


Fig. 1: Schiff bases derived from 2-amino-5-(substitutedphenyl) thiadiazole, substituted aromatic aldehyde and acetyl acetone

Characterization:

1. Characterization of 3-(Furan-2-yl methylene)-4-(5-p-tolyl-1, 3, 4 -2-ylimino) Pentane-2-one:
Yield: 82%, m. p.107 (0C), IR (KBr) ν cm $^{-1}$ 1708 (C=O), 1633 (C=N), 1020 (N-N), 634 (C-S-C), 1H NMR (400 M Hz, CDCl $_3$) δ 1.80 (S, 3H, N=C-CH $_3$), 2.10 (S, 3H, O=C-CH $_3$), 5.30 (S, 1H, HC=C), 2.41 (S, 3H, Ar-CH $_3$), 6.20-7.80 (M, 8H, Ar-H), Turbo spray MS- m/z 351 (M+1) $^+$, Anal. For C $_{19}H_{17}N_3O_2S$ (%): (C) 65.00, (H) 4.88, (N) 11.97, (S) 9.13, and (O) 9.12.

2. Characterization of (3E)-3-(4-Fluorobenzylidene)-4-(5-p-tolyl-

1,3,4-thiadiazole-2- ylimino) pentane-2-one:

Yield: 80%, m. p. 145 (0C), IR (KBr) ν cm $^{-1}$ 1693 (C=O), 1620 (C=N), 1018 (N-N), 632 (C-S-C), 1H NMR (400 M Hz, CDCl $_3$) δ 1.91 (S, 3H, N=C-CH $_3$), 2.25 (S, 3H, O=C-CH $_3$), 5.50 (S, 1H, HC=C), 6.30-8.35 (M, 8H, Ar-H), Turbo spray MS- m/z 379 (M+1) $^+$, Anal. For C $_{21}H_{18}N_3OSF$ (%): (C) 66.55, (H) 5.05, (N) 11.09, (S) 8.46, and (O) 8.44.

3. Characterization of (3E)-3-(4-methoxybenzylidene)-4-(5-p-tolyl-1,3,4 thiadiazole-2-ylimino)pentane-2-one:
Yield: 85%, m. p. 136 (0C), IR (KBr) ν cm $^{-1}$ 1693 (C=O), 1606 (C=N), 1033 (N-N), 634 (C-S-C), 1H NMR (400 M Hz, CDCl $_3$) δ 1.89 (S, 3H, N=C-CH $_3$), 2.15 (S, 3H, O=C-CH $_3$), 5.36 (S, 1H, HC=C), 6.89-7.79 (M, 8H, Ar-H), Turbo spray MS- m/z 391 (M+1) $^+$, Anal. For C $_{22}H_{21}N_3O_2S$ (%): (C) 67.40, (H) 5.05, (N) 10.71, (S) 8.17, and (O) 12.23.

4. Characterization of (3E)-3-(4-nitrobenzylidene)-4-(5-p-tolyl-1,3,4 thiadiazole-2-ylimino)pentane-2-one:
Yield: 81.5%, m. p. 138 (0C), IR (KBr) ν cm $^{-1}$ 1726 (C=O), 1662 (C=N), 1012 (N-N), 603 (C-S-C), 1H NMR (400 MHz, CDCl $_3$) δ 1.86 (S, 3H, N=C-CH $_3$), 2.19 (S, 3H, O=C-CH $_3$), 5.55 (S, 1H, HC=C), 7.35-8.23 (M, 8H, Ar-H), Turbo spray MS- m/z 406 (M+1) $^+$, Anal. For C $_{21}H_{18}N_4O_3S$ (%): (C) 62.12, (H) 4.72, (N) 13.80, (S) 7.89, and (O) 15.75.

5. Characterization of (3E)-3-(4-hydroxybenzylidene)-4-(5-p-tolyl-1,3,4 thiadiazole-2-ylimino)pentane-2-one: Yield: 86%, m. p. 132 (0C), IR (KBr) ν cm $^{-1}$ 1704 (C=O), 1635 (C=N), 1062 (N-N), 634 (C-S-C), 1H NMR (400 M Hz, CDCl $_3$) δ 1.84 (S, 3H, N=C-CH $_3$), 2.19 (S, 3H, O=C-CH $_3$), 5.55 (S, 1H, HC=C), 7.35-8.23 (M, 8H, Ar-H), Turbo spray MS- m/z 381 (M+1) $^+$, Anal. For C $_{21}H_{19}N_3O_2S$ (%): (C) 66.20, (H) 5.29, (N) 11.02, (S) 8.41, and (O) 12.59.

6. Characterization of (3E)-3-(4-chlorobenzylidene)-4-(5-p-tolyl-1,3,4 thiadiazole-2-ylimino)pentane-2-one:
Yield: 78.5%, m. p. 142 (0C), IR (KBr) ν cm $^{-1}$ 1707 (C=O), 1660 (C=N), 1012 (N-N), 632 (C-S-C), 1H NMR (400 MHz, CDCl $_3$) δ

1.90 (S, 3H, N=C-CH₃), 2.22 (S, 3H, O=C-CH₃), 5.45 (S, 1H, HC=C), 7.33-8.33 (M, 8H, Ar-H), Turbo spray MS- m/z 395 (M+1)⁺, Anal. For C₂₁H₁₈N₃OSCl (%): (C) 63.85, H 4.85, (N) 10.63, (S) 8.11, and (O) 8.10.

7. Characterization of (3E)-3-(4-methylbenzylidene)-4-(5-p-tolyl-1,3,4-thiadiazole-2-ylimino)pentane-2-one:

Yield: 79%, m. p. 139 (°C), IR (KBr) ν cm⁻¹ 1707 (C=O), 1656 (C=N), 1022 (N-N), 634 (C-S-C), ¹H NMR (400 M Hz, CDCl₃) δ 1.82 (S, 3H, N=C-CH₃), 2.17 (S, 3H, O=C-CH₃), 5.35 (S, 1H, HC=C), 6.90-7.85 (M, 8H, Ar-H), Turbo spray MS- m/z 375 (M+1)⁺, Anal. For C₂₂H₂₁N₃OS (%): (C) 70.46, (H) 5.91, (N) 11.20, (S) 8.55, and (O) 8.53.

8. Characterization of (3E)-3-(4-hydroxy-3-methoxybenzylidene)-4-(5-p-tolyl-1,3,4-thiadiazole-2-ylimino)pentane-2-one:

Yield: 87%, m. p. 144 (°C), IR (KBr) ν cm⁻¹ 1691 (C=O), 1633 (C=N), 1033 (N-N), 634 (C-S-C), ¹H NMR (400 M Hz, CDCl₃) δ 1.94 (S, 3H, N=C-CH₃), 2.22 (S, 3H, O=C-CH₃), 5.45 (S, 1H, HC=C), 6.82-7.79 (M, 8H, Ar-H), Turbo spray MS- m/z 407 (M+1)⁺, Anal. For C₂₂H₂₁N₃O₃S (%): (C) 64.92, (H) 5.45, (N) 10.32, (S) 7.88, and (O) 15.72.

A1

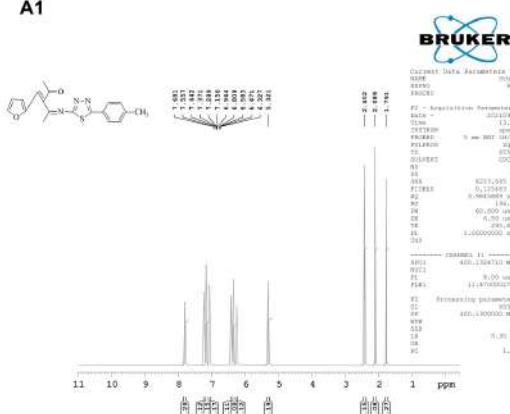


Fig. 2: ¹H NMR Spectra of ligand

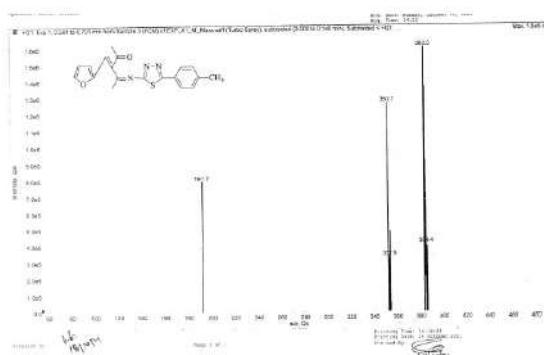


Fig 3 : Mass Spectra of ligand

Antimicrobial activity:

The *in-vitro* antibacterial (**Table 1**) and antifungal (**Table 2**) activities of synthesized Schiff bases have been studied by disc diffusion method^{34,35}. Four fungal and bacterial strains were tested for antifungal and antibacterial activity at concentrations of 100-200 g/mL in chloroform solvent. Minimum inhibitory concentration method was used to kill *Aspergillus niger*, *Candida albicans*, *Penicillium chrysogenum*, *Rhizopus spp.*, and *Staphylococcus aureus*, *Shigella spp.*, *Escherichia coli*, *Bacillus megaterium*. These bacterial and fungal cultures were cultured at 30°C for one day. Fluconazole and streptomycin, both antifungal and antibacterial, were employed. The diameter of the inhibitory zone was measured to evaluate activity (mm). It is observed that C₂₁H₁₉N₃O₂S and C₂₁H₁₈N₃OSCl ligands are more active against the fungal and bacterial strains *A. Nigar* and *S. aureus* as compare to other fungal and bacterial strains. Other ligands are moderately active against all fungal and bacterial strains^{36,37}. Antibacterial activities are shown in **Table 2**.

Conclusion

In summary, the author has synthesized new Schiff bases from 2-amino-5-(substituted phenyl)thiadiazole, substituted aromatic aldehyde and acetyl acetone using TBAHS catalyst in methanol as solvent at reflux temperature. The synthesized products were evaluated analytically, spectroscopically by elemental analysis FT-IR, ¹H NMR and mass spectra. All synthesized Schiff bases also screened by antimicrobial activity and found to be moderate to high antifungal and antibacterial activity with respect to standard.

TABLE 1 : PRELIMINARY IN VITRO ANTIBACTERIAL SCREENING ACTIVITIES OF LIGANDS

Compounds	Inhibition Zone (mm)							
	Antibacterial Activity		<i>Shigella</i> spp		<i>Escherichia coli</i>		<i>Bacillus megaterium</i>	
	100 (μ g/mL)	200 (μ g/mL)	100 (μ g/mL)	200 (μ g/mL)	100 (μ g/mL)	200 (μ g/mL)	100 (μ g/mL)	200 (μ g/mL)
C ₁₉ H ₁₇ N ₃ O ₂ S	+	+	+	+	+	+	+	+
C ₂₁ H ₁₈ N ₃ OSF	+	+	+	+	+	+	+	+
C ₂₂ H ₂₁ N ₃ O ₂ S	+	+	+	+	+	+	+	+
C ₂₁ H ₁₈ N ₄ O ₃ S	+	+	+	+	+	+	+	+
C ₂₁ H ₁₉ N ₃ O ₂ S	+	++	++	++	++	++	++	++
C ₂₁ H ₁₈ N ₃ OSCl	++	+++	++	+++	++	++	++	++
C ₂₂ H ₂₁ N ₃ OS	+	++	+	++	+	+	+	+
C ₂₂ H ₂₁ N ₃ O ₃ S	+	+	+	+	+	+	+	+

+ = 5-10 mm, ++ = 11-20mm, +++ = larger than 20mm and - = no inhibition.

TABLE 2 : PRILIMINARY IN VITRO ANTIFUNGAL SCREENING ACTIVITIES OF LIGANDS

Compounds	Inhibition Zone (mm)							
	Antifungal Activity		<i>Candida albicans</i>		<i>Penicillium chrysogenum</i>		<i>Rhizopus</i> spp	
	100 (μ g/mL)	200 (μ g/mL)	100 (μ g/mL)	200 (μ g/mL)	100 (μ g/mL)	200 (μ g/mL)	100 (μ g/mL)	200 (μ g/mL)
C ₁₉ H ₁₇ N ₃ O ₂ S	+	+	+	+	+	+	+	+
C ₂₁ H ₁₈ N ₃ OSF	+	+	+	+	+	+	+	+
C ₂₂ H ₂₁ N ₃ O ₂ S	+	+	+	+	+	+	+	+
C ₂₁ H ₁₈ N ₄ O ₃ S	+	+	+	+	+	+	+	++
C ₂₁ H ₁₉ N ₃ O ₂ S	++	++	++	++	++	++	++	++
C ₂₁ H ₁₈ N ₃ OSCl	++	++	++	++	++	++	++	+++
C ₂₂ H ₂₁ N ₃ OS	+	+	+	+	+	+	+	++
C ₂₂ H ₂₁ N ₃ O ₃ S	+	+	+	+	+	+	+	+

+ = 5-10 mm, ++ = 11-20mm, +++ = larger than 20mm and - = no inhibition.

Acknowledgements

Author thankful to Principal Dayanand Science College, Latur for providing laboratory facilities and also thankful to EmcurePharma for providing spectras.

Funding Details

Self-Funding

Conflict of interest

The author confirms that the content has no conflict of interest.

References

1. Jie, C.; Baohua. G.; Eugene, J.; LeBoeuf.; Hongjun, P.; Sheng. D. *Chemosphere*, **2002**, *48*, 59. doi:10.1016/s0045-6535(02)00041-3
2. Liu,M C.; Lin, T.S.; Sartorelli, A.C.J. *Med. Chem.*, **1992**, *35*, 3672. doi: 10.1021/jm00098a012, PMID 1433178.
3. Amer, S.; El-Wakiel, N.; El-Ghamry, H. *J. Mol. Struct.*, **2013**, *1049*,326. doi: 10.1016/j.molstruc.2013.06.059.
4. Neslihan, D.; Reyhan, U.; Ahmet, D. *Bioorg. Med. Chem.*, **2002**, *10*, 3717. doi: 10.1016/s0968-0896(02)00420-0, PMID 12413828.
5. Lei, S.; Hui-Ming, G.; Shu-Hua, T.; Huan-Qiu, Li.; Yong-Chun, S.; Hai-Liang, Z.; Ren-Xiang, T. *Eur. J. Med. Chem.*, **2007**, *42*, 558. doi: 10.1016/j.ejmech.2006.11.010, PMID 17194508.
6. Bernadette, S.; Brian, D.; Denise A, Egan.; Kevin, K.; Georgina, R.; Venkat, R.; Maureen, W. *Inorg. Chim. Acta*, **2010**, *363*, 4048. doi: 10.1016/j.ica.2010.08.009.
7. Abhay, N.;Netra, P.; Kiran, S. *Arab J Chem.*, **2016**, *9*, 48. doi: 10.1016/j.arabjc.2014.10.004.
8. Perumal, P.; Rajasree, R.; Gudaparthi, V.; Ekambaram H, Subramanian.; Seshaiah, K. *Eur. J. Med. Chem.*, **2005**, *40*, 225. doi: 10.1016/j.ejmech.2004.09.003, PMID 15694658.
9. OM, W.; Walsh, O.M.; Meegan, M. J.; Prendergast, R.M.; Nakib, T. A. *Eur. J. Med. Chem.*, **1996**, *31*, 989. doi: 10.1016/S0223-5234(97)86178-8.
10. Pandeya, S. N.; Sriram, D.; Nath, G.; DeClercq, E. *Eur. J. Pharmacol.*, **1999**, *9*, 25. doi: 10.1016/S0928-0987(99)00038-X.
11. Pandeya, S.N.; Sriram, D.; Nath, G.; DeClercq, E. *Pharm. Acta Helv.*, **1999**, *74*, 11. doi: 10.1016/s0031-6865(99)00010-2, PMID 10748620.
12. Yousif, E.; Majeed, A.; Al-Samarrae, K.;Salih,N.; Salimon, J.; Abdullah, B. *Arab J chem.*, **2013**, *10*, 1639.
13. Mohammad, N.U.; Didarul A, C.; Md, M.R.Am *J chemappl*, **2014**, *1*, 12.
14. Da Silva,C.M.; Da Silva,D.L.; Modolo, L.V.;Alves, R.B.; De Resende,M.A.; Martins,C.V.B.; De Fatima, A. *J. Adv. Res.*, **2011**, *2*, 1. doi: 10.1016/j.jare.2010.05.004.
15. Charef, N.; Sebti, F.; Arrar, L.; Djarmouni, M.; Boussoualim, N.; Baghiani, A.; Khennouf, S.; Ourari, A.; Aldamen, M.A.; Mubarak, M.S.; Peters, D.G. *Polyhedron*, **2015**, *85*, 450.
16. Bhusare, S.R.; Pawar, V.G.; Shinde, S.B.; Pawar, R.P.; Vibhute, Y.B. *Int. J. Chem. Sci.*, **2003**, *1*, 31. doi: 10.1016/j.poly.2014.09.006.
17. Hegazy, W.H.; Gaffar,A.E.D.M. *Am. Chem. Sci. J.*, **2012**, *2*, 86.
18. Osowole, A.A.; Oni, A.; Onyegbula, K.; Titilayo, H.A. *Int. Res. J. Pure Appl. Chem.*, **2012**, *2*, 211.
19. Zhang,B.; Haiqing, L.; Qinjuan, Xu.; Lirong, L.; B. Zhang, B. *Oncotarget*, **2017**, *8*, 13620. doi: 10.18632/oncotarget.14620, PMID 28099141.
20. Westheimer, F.H.; Taguchi, K.J. *Org. Chem.*, **1971**, *36*, 1570. doi: 10.1021/jo00810a033.
21. Chakraborti, Bhagat S.; Rudrawar, S. *TetrahedronLett.*, **2004**, *45*, 7641.
22. Dalpozzo, R.; De Nino,A.; Nardi, M.; Russo, B.A. *Procopio, Synthesis*, **2006**, *7*, 1127.

23. Naeimi, H.; Salimi, F.; Rabiei, K.J. *Mol. Catal. A Chem.*, **2006**, 260, 100. doi: [10.1016/j.molcata.2006.06.055](https://doi.org/10.1016/j.molcata.2006.06.055)
24. Barluenga, J.; Aznar, F.; Valdes, C. *Angew. Chem. Int. Ed.*, **2004**, 116, 347. doi: [10.1002/ange.200352808](https://doi.org/10.1002/ange.200352808)
25. Arluenga, J. B.; Jimenez-Aquino, A.; Fernandez, M.A.; Aznar, F.; Valdes, C. *Tetrahedron Lett.*, **2007**, 64, 778.
26. Jiang, L.; Jin, L.; Tian, H.; Yuan, X.; Yu, X.; Xu, Q. *Chem. Commun.*, **2011**, 47, 10833. doi: [10.1039/c1cc14242a](https://doi.org/10.1039/c1cc14242a)
27. Huang, B.; Tian, H.; Lin, S.; Xie, M.; Yu, X.; Xu, Q. *Tetrahedron Lett.*, **2013**, 54, 2861. doi: [10.1016/j.tetlet.2013.03.098](https://doi.org/10.1016/j.tetlet.2013.03.098)
28. Shiraishi, Y.; Ikeda, M.; Tsukamoto, D.; Tanaka, S.; Hirai, T. *Chem. Commun.*, **2011**, 47, 4811. doi: [10.1039/c0cc05615d](https://doi.org/10.1039/c0cc05615d), PMID 21416066.
29. Largeron, M.; Fleury, M.B. *Science*, **2013**, 339, 43. doi: [10.1126/science.1232220](https://doi.org/10.1126/science.1232220), PMID 23288531
30. Lan, Y.S.; Liao, B.S.; Liu, Y.H.; Peng, S.M.; Liu, S.T. *Eur. J. Org. Chem.*, **2013**, 2013, 5160. doi: [10.1002/ejoc.201300507](https://doi.org/10.1002/ejoc.201300507)
31. Largeron, M. *Eur. J. Org. Chem.*, **2013**, 2013, 5225. doi: [10.1002/ejoc.201300315](https://doi.org/10.1002/ejoc.201300315)
32. Joshia, H.; Kamounahb, F.S.; Gooijera, C.; Zwana, G.; Antonovc, L. *J. Photochem. Photobiol., B*, **2002**, 152, 183. doi: [10.1016/S1010-6030\(02\)00155-7](https://doi.org/10.1016/S1010-6030(02)00155-7).
33. Kasimala, M. *Caribbean J. Sci. Tech.* **2021**, 9(1), 7-9. doi: <https://doi.org/10.55434/CBI.2021.9106>
34. Ramya Krishna Pallapati, Baby Ramana Mutchu, Bala Murali Krishna Khandapu, Umamaheswara Rao Vanga, Ravi Varala, Hari Babu Bollikolla. *Chemistry Africa*, **2020**, 3(1), 881-888. doi: [10.1007/s42250-020-00184-x](https://doi.org/10.1007/s42250-020-00184-x)
35. Joga Rao, Y.S.V.S.; Lanka, A.; Bollikolla, H.; Ramachandran, D. *Caribbean J. Sci. Tech.* **2019**, 7(1), 058-064. doi: <https://doi.org/10.55434/CBI.2019.7109>.
36. Kim, H.C.; Lee, H.K.; Choi, A.; Jeon, S.W. *Bull. Korean Chem. Soc.*, **2007**, 28, 538. doi: [10.5012/bkcs.2007.28.4.538](https://doi.org/10.5012/bkcs.2007.28.4.538).
37. Andreoli, P.; Cainelli, G.; Giacomini, D.; Martelli, G.; Panunzio, M. *Tetrahedron Lett.*, **1986**, 27, 1695. doi: [10.1016/S0040-4039\(00\)84350-6](https://doi.org/10.1016/S0040-4039(00)84350-6).